

Research Title Chemistry of Some Organotin Compounds
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Abstract

Tetraorganotins of $\text{Sn}(\text{C}_2\text{H}_5)_4$ and $\text{Sn}(\text{C}_6\text{H}_5)_4$ have been prepared by Grignard reagent and Wurtz reaction respectively. They were later used as starting materials to prepare their derivatives. Adduct formations with donor ligands such as dimethyl sulfoxide, pyridine, phenanthroline and acetylacetone yielded compounds with octahedral structures. Various techniques such as infrared, melting point and chemical analysis were used to determine the possible structures of the compounds. Infrared spectroscopy indicated that the alkyl groups were coordinated to Sn(IV). By using the data from melting point and chemical analysis it was suggested that organotins $\text{Sn}(\text{C}_2\text{H}_5)_4$, $\text{Sn}(\text{C}_2\text{H}_5)_2\text{Cl}_2$, $\text{Sn}(\text{C}_6\text{H}_5)_4$, $\text{Sn}(\text{C}_6\text{H}_5)_3\text{Cl}$ were tetrahedral and $\text{Sn}(\text{C}_2\text{H}_5)_2\text{Cl}_2 \cdot 2\text{DMSO}$, $\text{Sn}(\text{C}_2\text{H}_5)_2\text{Cl}_2 \cdot 2\text{Py}$, $\text{SnCl}_4 \cdot 2\text{DMSO}$, $\text{SnCl}_4 \cdot 2\text{Py}$, $\text{SnCl}_4 \cdot \text{Phen}$, $\text{SnCl}_2(\text{C}_5\text{H}_7\text{O}_2)_2$ were octahedral.